Regional geochemical variations in a metamorphosed black shale: a reconnaissance study of the Silurian Smalls Falls Formation, Maine, USA

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ABSTRACT

A reconnaissance geochemical study of 21 samples of sulphidic black phyllite and schist from the Silurian Smalls Falls Formation in Maine was undertaken in order to evaluate compositional changes during regional metamorphism. These samples represent variably metamorphosed black shale. Analyzed samples come from the chlorite zone in northern Maine and the biotite, garnet, and staurolite-andalusite zones in west-central Maine. Strata of the Smalls Falls Formation are distinctive in containing abundant pyrite and/or pyrrhotite (total S =1.2-9.7 wt %), but only minor organic matter or graphite (TOC = 0.43-1.85 wt %); TOC/S ratios are uniformly low (average = 0.37 ± 0.22). Median enrichment factors were calculated for each element by normalizing the concentration to Ti in each sample to the Ti-normalized median composition of global black shale. In the chlorite zone, moderate to large decreases in enrichment factors (-23.1 to -49.8%) are evident for V, Cr, Cu, Ni, Zn, Pb, Sb, and U, attributed here to various factors during sedimentation plus variable element mobility during diagenesis. With increasing metamorphic grade (biotite through staurolite-andalusite zones), systematic small to extreme decreases (-14.5 to -99.0%) were found for Ba, Sb, Au, and U, together with less-systematic moderate to large decreases (-35.4 to -61.1%) for V and As. Molybdenum shows an extreme decrease (-94.7%) from the garnet to staurolite-andalusite zones. Excluding Ba, these results are interpreted to mainly reflect mobility of trace elements during pyrite recrystallization, and during the metamorphic transformations of organic matter to graphite and of pyrite to pyrrhotite. Moderate to large increases for Rb (+28.1 to +61.5%) and Th (+39.1 to +47.3%) from the biotite to staurolite-andalusite zones likely record the introduction of alkalis and mass loss, respectively, during metamorphism. Three samples from one site in the garnet zone differ in having anomalously high Fe/Al and low La/Yb ratios, attributed here to epigenetic formation of pyrite and related leaching of light rare earth elements during syn-metamorphic, channelized fluid flow.

Geologic and geochemical data indicate that strata of the Smalls Falls Formation were deposited during an interval of anoxia on the northwestern flank of the Central Maine Basin, for which detrital sources included an evolved continental arc. Onset of anoxia coincided with deposition of the Mayflower Hill Formation of the Vassalboro Group, on the basin's southeastern flank, related to emergence of the Brunswick subduction complex. We suggest that this emergence played a role in promoting both lateral and vertical circulation changes, nutrient loading, and deoxygenation through subsequent basin closure that culminated with Acadian deformation and metamorphism. Based on the relatively high contents of total sulphur present in our Smalls Falls samples, sediments in the Black Sea represent the only known plausible candidate among those in modern suboxic to euxinic basins.

RÉSUMÉ

Une étude géochimique de reconnaissance de 21 échantillons de schiste et de phyllite noirs sulfurés de la Formation silurienne de Smalls Falls dans le Maine a été réalisée pour évaluer les changements de la composition survenue durant le métamorphisme régional. Les échantillons en question représentent du schiste noir métamorphisé de façons variables. Les échantillons analysés proviennent de la zone de chlorite dans le nord du Maine et des zones de biotite, de grenat et de staurolite-andalousite dans le centre-ouest du Maine. Les strates de la Formation de Smalls Falls se distinguent par leur abondance en pyrite ou en pyrrhotite (S total = 1,2 à

9,7 % en poids), mais par une quantité modeste seulement de matière organique ou de graphite (COT = 0,43 à 1,85 % en poids); les ratios de COT/S sont uniformément bas (moyenne = 0.37 ± 0.22). On a calculé les facteurs d'enrichissement médians de chaque élément en normalisant la concentration en fonction du TI dans chaque échantillon d'après la composition médiane normalisée en TI de l'ensemble du schiste noir. Dans la zone de chlorite, des réductions moyennes à marquées des facteurs d'enrichissement (-23,1 à -49,8 %) sont évidentes dans le cas du V, du Cr, du Cu, du Ni, du Zn, du Pb, du Sb et de l'U; elles sont ici attribuées à divers facteurs ayant agi durant la sédimentation ainsi qu'à la mobilité variable des éléments durant la diagenèse. Au fur et à mesure que s'accentue l'intensité du métamorphisme (des zones de biotite à celles de staurolite-andalousite), on relève des réductions minimes à extrêmes systématiques (-14,5 à -99,0 %) du Ba, du Sb, de l'Au et de l'U, de même que des réductions moyennes à marquées moins systématiques (-35,4 à -61,1 %) du V et de l'As. Le molybdène accuse une diminution extrême (-94,7 %) dans les zones de grenat passant à de la staurolite-andalousite. Sauf pour ce qui est du Ba, ces résultats sont principalement interprétés comme une réflexion de la mobilité des éléments traces durant la recristallisation de la pyrite et durant les transformations métamorphiques de la matière organique en graphite et en pyrite, puis en pyrrhotite. Les augmentations moyennes à prononcées du Rb (+28,1 à +61,5 %) et du Th (+39,1 à +47,3 %) dans les zones de biotite devenant de la staurolite-andalousite témoignent vraisemblablement de l'introduction de minéraux alcalins et d'une perte massique, respectivement, durant le métamorphisme. Trois échantillons provenant d'un emplacement à l'intérieur de la zone de grenat se distinguent par des ratios anormalement élevés de Fe/Al et faibles de La/Yb, attribués dans ce cas à la formation épigénétique de pyrite et au lessivage d'éléments de terres rares légers durant un écoulement des fluides canalisé, synmétamorphique. Des données géologiques et géochimiques révèlent que les strates de la Formation de Smalls Falls se sont déposées au cours d'un intervalle d'anoxie sur le flanc nord-ouest du bassin central du Maine, dont les sources détritiques ont compris un arc continental évolué. Le début de l'anoxie a coïncidé avec le dépôt de la Formation de Mayflower Hill du groupe de Vassalboro sur le flanc sud-est du bassin, conjointement à l'émergence du complexe de subduction Brunswick. Nous croyons que cette émergence a contribué aux changements de la circulation latéraux et verticaux, à la charge en éléments nutritifs et à la désoxygénation par la fermeture subséquente du bassin, qui a culminé avec le métamorphisme et la déformation acadiens. À en juger d'après la teneur relativement élevée en soufre total présente dans nos échantillons de Smalls Falls, les sédiments de la mer Noire représentent le seul candidat plausible connu parmi ceux des bassins suboxiques à euxiniques modernes.

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INTRODUCTION

The formation of organic-rich sedimentary rocks typically reflects several factors including high biological productivity, a slow sedimentation rate, and low redox states of bottom waters and pore fluids (Wignall 1994, and references therein). Many studies have evaluated compositional changes in organic-poor shale as a function of diagenesis and metamorphism (e.g., Ague 1991, 1994a; Land *et al.* 1997; Milliken 2003; Masters and Ague 2005; Merriman *et al.* 2009), and several have discussed diagenetic effects in organic-rich shale (Hannigan and Basu 1998; Lev *et al.* 2008; Abanda and Hannigan 2006; Yang *et al.* 2018a), but to our knowledge no publications exist that evaluate the influence of metamorphism on the full spectrum of inorganic components in black shales.

This report presents whole-rock analyses for major elements, trace elements, and rare earth elements (REE) in a reconnaissance study (21 samples) of variably sulphidic black phyllite and schist from the Silurian Smalls Falls Formation in northern and western Maine. These samples represent variably metamorphosed black shale. The geochemical data reported here are used as a basis for better understanding the mineralogical and compositional effects of Silurian diagenesis and Devonian regional metamorphism. We also present trace element data bearing on the provenance of the original sediments of this formation and related tectonic implications. Early geochemical studies of the Smalls Falls Formation were restricted to analysis of only a few samples (Cullers *et al.* 1997). Two field trip guidebooks by Guidotti and Van Baalen (2001) and Van Baalen (2006) presented results for a limited number of trace elements, together with a discussion of environmental issues focused on arsenic. Our paper reports data for a much larger suite of elements with greater analytical precision, and supersedes the bulk compositions reported in these two guidebooks as well as those in the guidebook chapter by Van Baalen *et al.* (2017).

GEOLOGIC SETTING

The Smalls Falls Formation forms part of the Central Maine Basin (CMB; Fig. 1) that contains marine metasedimentary rocks of Late Ordovician to Early Devonian age (Osberg *et al.* 1985; Ludman *et al.* 2017). On its northwest flank, strata of the CMB conformably overlie Ordovician and Cambrian marine sedimentary and volcanic rocks of the Bronson Hill belt (Moench 2006). Relationships along the southeastern margin of the CMB are diverse, ranging from fault contacts in southern Maine (e.g., Liberty–Orrington belt) to a Salinic (Silurian) unconformity in New Brunswick (Ludman *et al.* 2017). Paleocurrent data and stratigraphic relationships suggest that the CMB received sediment from the northwest during the Early Silurian, and subsequently also from the southeast (Bradley and Hanson 2002).



Figure 1. Simplified geological map of Maine showing distribution of the Smalls Falls Formation and correlatives, regional metamorphic zones, and sample locations. Plutonic rocks consist of granite with subordinate gabbro, chiefly of Ordovician and Devonian age. Uncolored areas are mainly lower to middle Paleozoic metasedimentary and metavolcanic rocks. Geology and metamorphic zones adapted from Osberg *et al.* (1985). Regional geology for location 10 is available in Roy (1981) and Roy *et al.* (1983); for locations 11–14, see Moench and Pankiwskyj (1988).

Strata of the Smalls Falls Formation and correlatives extend ca. 600 km from northern Maine southward through New Hampshire and Massachusetts. In west-central Maine, the Smalls Falls Formation overlies, with sharp contact, quartz-rich turbidites of the Silurian Perry Mountain Formation and sandstone and conglomerate of the underlying Rangeley Formation (Boone et al. 1970, Boone 1973; Moench 2006). The Smalls Falls underlies calcareous turbidites of the Silurian-Devonian Madrid Formation and predominantly siliciclastic strata of the Devonian Seboomook Group. In northern and northeastern Maine, the correlative upper slate member ("sulfidic phyllite") of the Allsbury Formation overlies phyllite, siltstone, and sandstone of the lower part of the Allsbury, and underlies quartzo-feldspathic greywacke and minor non-graphitic slate/phyllite of the Silurian Lawler Ridge Formation correlated with the Madrid Formation,

and turbidites of the Devonian Seboomook Group (Roy *et al.* 1983). According to Ludman *et al.* (2017), strata of the Smalls Falls Formation record the last pre-orogenic sedimentation prior to development of the Acadian foreland basin during the early Emsian (e.g., Bradley *et al.* 2000).

The Smalls Falls Formation was assigned by Moench *et al.* (1995) to a Ludlovian (Late Silurian) age based on rare graptolites in the correlated Parkman Hill Formation several tens of kilometres east of the Smalls Falls type locality. More recently, Bradley and O'Sullivan (2016) reported a detrital zircon date of 409 Ma for the youngest age population in the Smalls Falls Formation, suggesting a maximum depositional age of Pragian (Early Devonian). However, this maximum age may underestimate the true depositional age because of possible inadvertent measurement of high-U metamorphic rims on the zircon grains, as well as other geologic factors

(see Bradley and O'Sullivan 2016). For the purposes of this report, we assume that the Smalls Falls Formation is Ludlovian pending more detailed studies.

Thicknesses of Smalls Falls strata and correlatives vary from ca. 115 m in the northeast (Roy et al. 1983), to ca. 480 m in the central part of the study area (Boone 1973), to approximately 1150 m in western Maine (Moench 2006). The formation thins southeastward from its type locality on the northwest flank of the Bear Hill syncline across the Strickland Hill anticline to the Currier Hill syncline (Moench and Pankiwskyj 1988). Farther southeast, the Smalls Falls is absent, and instead sandstones of the time-equivalent Mayflower Hill Formation conformably overlie the Wenlock Waterville Formation in an overall coarsening-upwards sequence (Marvinney et al. 2010). In addition to black phyllite or schist, interbeds of sulphidic metasiltstone or quartzite up to several centimeters thick are common in many exposures. In the type area at Smalls Falls, quartzite-rich turbidites typically display grading and cross laminations; feldspathic and calcareous metasiltstone (or calc-silicate) beds are common in the upper part of the formation (Moench 2006). No evidence was discovered in our sampled outcrops of veins or of other epigenetic hydrothermal products, although surface oxidation in many places prevents a thorough evaluation of this possibility. Importantly, on the scale of hand specimens (and drill cores), no quartz- or sulphide-rich veins were found.

METAMORPHIC FRAMEWORK

Metamorphic zones vary greatly in Maine (Fig. 1). In northernmost exposures of the Smalls Falls Formation, sedimentary strata are at prehnite-pumpellyite grade (Richter and Roy 1974). Progressively to the southwest, the metamorphic grade increases through lower greenschist (chlorite, biotite), to upper greenschist (garnet), middle amphibolite (staurolite-andalusite), and upper amphibolite (sillimanite and sillimanite-K-feldspar). Regional metamorphic assemblages of the greenschist-facies event (M1) were overprinted by assemblages produced during three later events that developed within the thermal regimes of high temperature-low pressure (HT-LP) associated with intrusion of granitic and subordinate gabbroic plutons of Devonian (M2, M3) and Carboniferous (M4) age (Guidotti and Holdaway 1993; Guidotti et al. 1996; Guidotti and Johnson 2002). In western Maine, proximity and annular distribution of the biotite, garnet, staurolite-andalusite, sillimanite, and sillimanite-K-feldspar zones around plutons (Boone 1973; Guidotti 1974; Moench and Pankiwskyj 1988; Guidotti et al. 1996) suggests a genetic link, as proposed also for the type area of Buchan HT-LP metamorphism in northeastern Scotland (e.g., Lyubetskaya and Ague 2010).

Iron-rich silicate and oxide minerals including chloritoid, garnet, staurolite, and ilmenite are rare to absent in sulphide-rich samples from the Smalls Falls Formation owing to sequestering of Fe²⁺ via metamorphic sulphidation reactions during the formation of pyrrhotite from pyrite. Instead, among silicate minerals, Mg-rich chlorite, phlogopite, and cordierite predominate, and rutile among oxide minerals, depending on metamorphic grade (Henry 1981; Guidotti *et al.* 1977; Guidotti and Van Baalen 2001; Van Baalen 2006).

METHODS

Samples were collected from outcrops and road cuts mainly with a small diamond drill in order to obtain unweathered rock. Studied drill cores, 2.5 cm in diameter, are mostly 8 to 12 cm long. Sample locations and descriptions are given in Table 1. Geological settings for the sampled outcrops are available in Roy et al. (1983) for the chlorite zone, Ludman (1977) for the biotite zone, Boone (1973) for the garnet zone, and Guidotti and Van Baalen (2001) for the staurolite-andalusite zone. Samples analyzed in our study come from the M1 chlorite zone, and the M2 biotite, garnet, and staurolite-andalusite zones (Fig. 1). In addition to these four metamorphic zones, attempts were made to collect samples of sulphide-rich black shale from the prehnite-pumpellyite zone and sulphide-rich schist from the sillimanite and sillimanite + K-feldspar zones, but without success. Sampling was focused on quartz-poor intervals, but in some cases it was not possible to avoid the inclusion of thin (<5 mm) laminae of metasiltstone or quartzite in the samples.

Major elements were analyzed by X-ray fluorescence at AGAT Laboratories. Total sulphur was determined by an automated LECO analyzer and carbonate carbon by coulometric titration, both at AGAT Laboratories. Trace elements and rare earth elements (REE) were determined at SGS Laboratories, the former by Inductively Coupled Plasma-Atomic Emission Spectrometry (ICP-AES) and the latter by Inductively Coupled Plasma-Mass Spectrometry (ICP-MS) after sodium peroxide fusion. Contents of Au, Pd, and Pt were obtained by fire assay. Four USGS rock standards including shales SCo-1, SBC-1, and SGR-1, and black shale SDO-1, were run to monitor accuracy and precision. Details of method descriptions and analytical performance are presented in Granitto et al. (2017). Complete analyses and related element ratios are reported in Tables A1 and A2 in the Appendix.

Mineralogical studies were carried out at the U.S. Geological Survey in Reston, Virginia, using standard petrography (transmitted and reflected light microscopy) and scanning electron microscopy (SEM). Imaging and identification of fine-grained minerals were done with a Hitachi SU-5000 field-emission SEM (15 kV, spot intensity 30, working distance 10 mm), coupled to an EDAX Octane Plus Silicon Drift Detector for compositional analysis via Energy Dispersive Spectrometry (EDS).

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Location*	Sample number	Latitude and longitude	Metamorphic grade	Description	Notes
10	JS-13-10A	45°45′05″N, 68°27′19″W	Chlorite	Black pyritic phyllite; orange carb	Strat middle of ocp
10	JS-13-10B	45°45′05″N, 68°27′19″W	Chlorite	Black pyritic phyllite; slt laminae	2 m strat below A
10	JS-13-10C	45°45′05″N, 68°27′19″W	Chlorite	Black pyritic phyllite; slt laminae	2 m strat below B
10	JS-13-10D	45°45′05″N, 68°27′19″W	Chlorite	Dark grey pyritic phyllite; slt laminae	2 m strat below first ss bed
10	JS-13-10E	45°45′05″N, 68°27′19″W	Chlorite	Dark grey pyritic phyllite	2 m strat above first ss bed
11	JS-13-11A	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist	Strat highest exposure
11	JS-13-11B	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist	0.5 m strat below A
11	JS-13-11C	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist; slt laminae	1 m strat below B
11	JS-13-11D	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist; slt laminae	0.5 m strat below C
11	JS-13-11E	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist	1 m strat below D
11	JS-13-11F	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist	0.5 m strat below E
11	JS-13-11G	44°46′17″N, 69°42′31″W	Biotite	Dark grey pyritic schist	0.1 m strat below E
12	JS-13-12A	45°03′07″N, 69°53′13″W	Garnet	Dark grey pyrite-rich schist	South end of ocp
12	JS-13-12B	45°03′07″N, 69°53′13″W	Garnet	Dark grey pyrite-rich schist	ca. 8 m strat below A
12	JS-13-12C	45°03′07″N, 69°53′13″W	Garnet	Dark grey pyrite-rich schist	ca. 8 m strat below B
13	JS-13-13B	45°04′16″N, 69°54′53″W	Garnet	Dark grey pyrite-rich schist	Locally abundant Py
14	JS-13-14A	44°51′33″N, 70°31′02″W	Staurolite-andalusite	Dark grey pyrrhotite-rich schist	Strat upper part of ocp
14	JS-13-14B	44°51′33″N, 70°31′02″W	Staurolite-andalusite	Dark grey pyrrhotite-rich schist	ca. 7 m strat below A
14	JS-13-14C	44°51′34″N, 70°31′04″W	Staurolite-andalusite	Dark grey pyrrhotite-rich schist	ca. 30 m strat below B
14	JS-13-14D	44°51′48″N, 70°31′14″W	Staurolite-andalusite	Dark grey pyrrhotite-rich schist	Abundant coarse andalusite
14	JS-13-14E	44°51′48″N, 70°31′15″W	Staurolite-andalusite	Dark grey pyrrhotite-rich schist	Abundant coarse andalusite

*Geographic locations: 10, Benedicta, outcrops on W side of southbound Interstate I-95; 11, Skowhegan, outcrop along N shore of Kennebec River ~100 m south of Rte 2; 12, Bingham, outcrop at W end of Bridge Street on W side of Kennebec River; 13, Outcrop along Pleasant Ridge Rd north of Wyman Dam; 14, Roadcuts along Rte 4 N of type area at Smalls Falls.

Abbreviations: carb, carbonate; ocp, outcrop; Po, pyrrhotite; Py, pyrite; ss, metasandstone; slt, metasiltstone; strat, stratigraphic or stratigraphically.

RESULTS

Mineralogy

Mineral assemblages vary partly as a function of metamorphic grade. In the chlorite zone, assemblages comprise quartz, muscovite, and chlorite with minor plagioclase and pyrite, and sparse Fe-Mg carbonate and apatite. Carbon phases, which include organic matter (OM) and graphite, are dispersed among other constituents. Pyrite typically forms subhedral and euhedral grains mainly in silty laminae (Fig. 2a). Less common is framboidal pyrite, uniformly less than 7 μ m in diameter, occurring preferentially in a local matrix of Fe-Mg carbonate and albite, or quartz (Fig. 3a, b). Monazite, ilmenite, and zircon are trace constituents.

Samples from the biotite zone are also dominated by quartz and muscovite, with minor plagioclase and pyrite, local phlogopite, and sparse apatite and ilmenite (Fig. 2b). Carbonate is nearly absent and carbon phases are less abundant than in samples from the chlorite zone. Significantly, pyrite occurs only as subhedral or euhedral crystals, locally with inclusions of chalcopyrite or galena. Trace minerals include monazite and zircon.

Garnet-zone samples comprise quartz and muscovite and, with one exception, lack carbonate. Garnet is absent

from sulphide-rich samples. Plagioclase is present in a single sample. Apatite, ilmenite, and carbon phases are sparse constituents; monazite and zircon occur in trace amounts. Pyrite forms anhedral grains aligned in the dominant foliation (Fig. 2c), and subhedral to euhedral crystals with large pyrrhotite and small sphalerite inclusions (Fig. 2d).

Samples from the staurolite-andalusite zone mostly contain quartz and muscovite with minor phlogopite and coarse-grained (up to several cm) iron sulphides, and sparse plagioclase. Cordierite and andalusite are present locally. Carbon phases are negligible; apatite and rutile are rare. Monazite and zircon are trace constituents. Pyrrhotite predominates over pyrite, the former locally containing inclusions and veinlets of chalcopyrite (Fig. 2e), and inclusions of galena (Fig. 2f) and/or sphalerite. Staurolite is absent in sulphide-rich samples.

Yttrobetafite-(Y) $[(Y,U,Ce)_2(Ti,Nb,Ta)_2O_6(OH)]$ was discovered in one of our sampled outcrops by Van Baalen *et al.* (2005). Although this mineral was not found during our reconnaissance SEM work, it is likely present in many samples, based on whole-rock geochemical data that show a strong positive correlation (R² = 0.73) of Y+Ce+U with Ti+Nb+Ta.



Figure 2. Photomicrographs in reflected light of analyzed samples from the Smalls Falls Formation. (a), Subhedral pyrite with silicate inclusions and thin oxidized rim, in matrix of silicates, minor carbonate, and trace sulphide; sample JS-13-10B (chlorite zone). (b), Euhedral pyrite in matrix of silicates and minor carbonate with sparse ilmenite; sample JS-13-11B (biotite zone). (c), Anhedral foliated pyrite in silicate matrix; sample JS-13-12A (garnet zone). (d), Large pyrrhotite and small sphalerite inclusions in subhedral pyrite; sample JS-13-12B (garnet zone). (e), Anhedral pyrrhotite with rims and veinlets of chalcopyrite; sample JS-13-14A (staurolite-andalusite zone). (f), Anhedral pyrrhotite with inclusion of galena; sample JS-13-14B (staurolite-andalusite zone). Abbreviations: Cp, chalcopyrite; Gl, galena; Ilm, ilmenite; Po, pyrrhotite; Py, pyrite; Sl, sphalerite.

Geochemical data

Whole-rock analyses for major, minor, and trace elements (including REE) were acquired on 21 samples of black phyllite and schist from the Smalls Falls Formation. Significant variations among major elements are shown by SiO₂ (54.6–72.0 wt %), Al₂O₃ (7.15–20.2 wt %), Fe₂O₃^T (2.30–12.7 wt %), MgO (0.93–3.53 wt %), CaO (0.43–3.75 wt %), Na₂O (0.43–1.82 wt %), and K₂O (1.41–4.34 wt %). Contents of MnO and P₂O₅ are low (<0.32 and <0.20 wt %, respectively). TiO₂ varies from 0.35 to 0.97 wt %. TOC values (organic + graphitic

C) are 0.43 to 1.85 wt %; most samples have below 1.0 wt %. Significantly, total sulphur is generally high (1.2–9.7 wt %); these total S contents wholly reflect sulphide sulphur (pyrite \pm pyrrhotite \pm chalcopyrite \pm sphalerite \pm galena), given the absence of barite or appreciable amounts of other potentially sulphur-rich minerals such as apatite. Concentrations of base and related transition metals include Co = 12.6 to 38.6 ppm, Cu = 24 to 64 ppm, Ni = 34 to 82 ppm, Pb = 11 to 34 ppm, and Zn = 6 to 132 ppm. Contents of Cr and V are ≤ 100 and ≤ 217 ppm, respectively. Data for high field strength elements show the following ranges: Nb (7–18 ppm),



Figure 3. SEM-BSE images of framboidal pyrite in chlorite-grade rock from the Smalls Falls Formation. (a), Small framboids in matrix of Fe-Mg carbonate and albite; sample JS-13-10B. (b), Small framboids in matrix of albite and muscovite; sample JS-13-10B. Note diameters of framboidal pyrite grains are all <8 μ m (see text). Abbreviations: Ab, albite; Cb, carbonate; Ms, muscovite; Qtz, quartz; Py, pyrite.

Th (4.20–17.4 ppm), and Zr (97.3–331 ppm). Uranium concentrations vary from 2.87 to 15.2 ppm. REE contents vary widely (e.g., La = 11.7–48.9 ppm; Yb = 1.50–5.30 ppm). Similar ranges in concentration were reported by Van Baalen (2006) for a different suite of six samples from the Smalls Falls Formation, from different metamorphic grades, including for Ni (<100 ppm), Pb (<21 ppm), Cr (<155 ppm), V (<167 ppm), and U (\leq 10 ppm).

Carbon-sulphur-molybdenum relationships

Figure 4a shows a plot of whole-rock data for total TOC vs total sulphur in which TOC includes both carbonaceous and graphitic carbon. Noteworthy are the uniformly low TOC contents (most <1.0 wt %) and generally high total S values (most ~2–4 wt %; up to 9.7 wt %). Ratios of TOC/ total S are all less than 1.0, averaging 0.37 ± 0.22 . Data for the Smalls Falls samples differ greatly from those for modern marine sediments, euxinic Black Sea sediments, and

sulphide-rich black shales of the Upper Devonian–Lower Mississippian Bakken Formation (Montana) and the Jurassic Jet Rock Member of the Whitby Mudstone Formation (U.K.). To our knowledge, no modern or ancient unmineralized black shales or their lithologic equivalents have such low TOC yet high total S contents above ca. 2 wt %.

A plot of TOC vs Mo (Fig. 4b) shows a similar trend in which uniformly low TOC contents are paired by relatively low Mo (\leq 61 ppm). Compared to modern anoxic to euxinic settings, data for the Smalls Falls samples display much lower Mo/TOC ratios. Moreover, none of our samples has Mo contents at or above 100 ppm, which is the inferred lower limit for persistent euxinic (sulphidic) conditions during deposition (Scott and Lyons 2012).

Figure 4c is a plot of total S vs Mo. These data reinforce the key point that despite high S contents that approach 10 wt %, Mo concentrations are uniformly low. For comparison, most analyses of the locally pyrite-rich Bakken Formation (Scott *et al.* 2017b) show higher Mo at a given total S content than our samples from the Smalls Falls Formation. Relative to the Bakken data, Smalls Falls samples that have more than 4 wt % total S are depleted in Mo by at least 20 to 70 ppm.

Comparisons to median black shale

The evaluation of geochemical changes in black shale during diagenesis and low-grade metamorphism requires data for unmetamorphosed samples for use as a baseline composition. However, no such samples were available for this study. As a result, we compare our analyses for the Smalls Falls Formation to a global database of black shale geochemistry (Ketris and Yudovich 2009), and emphasize that the latter is not considered a premetamorphic equivalent of Smalls Falls samples. Note that this global database does not include highly metalliferous black shales (cf. Coveney 2003; Johnson *et al.* 2017).

Table 2 presents basic statistics for trace elements and REE, together with the global data for median black shale as compiled by Ketris and Yudovich (2009). Data for average shale (Li and Schoonmaker 2003) are shown for comparison. Also listed are median enrichment factors and percent changes calculated in relation to the global median black shale composition. Compared to this black shale median, Th and Rb in Smalls Falls samples show moderate positive changes of +39.6 and +41.0%, respectively. Moderate to very large negative changes (-30.0 to -94.6%) are evident for Ge, V, Cr, Cu, Ni, Cd, Zn, Pb, Au, Bi, Tl, Sb, As, Sn, and U. Other elements display variable changes in the range of ± 10 to $\pm 30\%$. Changes of less than 10% are not considered significant, given the small database available for this study. Note that the results for some trace elements are not presented due to the limited number of samples (n = <700) reported by Ketris and Yudovich (2009), compared to those for most other trace elements listed in their database (n > 1000; many >5000).

Rare earth element patterns

Abundances of REE normalized to average Post Archean Australian Shale (PAAS; Taylor and McLennan 1985) range from ca. 0.3 to 2.0 (Fig. 5). Samples from the chlorite and biotite zones display broadly flat patterns with greater total abundances in the former group of samples, and a narrow range of La/Yb ratios (10.4–14.1; Appendix). The middle rare earth elements (MREE) show slight enrichment, relative to PAAS. Three of four garnet-zone samples display significant depletion of light rare earth elements (LREE) with La/Yb ratios of 4.7 to 7.7. In the staurolite-andalusite zone, La/Yb ratios range from 8.9 to 14.5; one sample has enriched heavy



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rare earth elements (HREE) compared to samples from the chlorite and biotite zones. No significant Ce anomalies exist, based on calculated Ce/Ce* values (PAAS basis) that range from 0.95 to 1.01. Most samples have Eu/Eu* values (PAAS basis) of 0.90 to 1.09 that are neither significantly negative nor positive, but nine samples have small positive anomalies (Eu/Eu* = 1.10-1.45); the two highest Eu/Eu* values of 1.23 and 1.45 occur in the biotite and garnet zones, respectively. Small positive Tm anomalies present in all samples are likely analytical artefacts (e.g., Losh and Rague 2018) and are not discussed further.

DISCUSSION

Provenance

Strata of the Smalls Falls Formation originated as organicrich clastic sediments that were deposited in a marine basin whose source region lay chiefly to the northwest (present-day coordinates) based on the paleocurrent data of Bradley and Hanson (2002). More insight into provenance of the original sediments comes from immobile trace element ratios, which were likely unaffected by diagenesis and metamorphism thus supporting their use in comparing data for the metamorphic rocks of the Smalls Falls Formation to those of unmetamorphosed sediments. Figure 6 indicates that, with one exception, all samples have Th/Sc ratios >0.5, which suggests predominantly felsic sources; the trend and range of Zr/Sc ratios is consistent with appreciable sediment recycling and zircon concentration (McLennan et al. 1993). This interpretation is supported by the occurrence of predominantly Proterozoic grains in the detrital zircon age spectrum for the Smalls Falls Formation reported by Bradley and O'Sullivan (2016). Use of Sc in such plots may be questioned if this element and related REE were mobile during metamorphism (e.g., Ague 2017); however, strong

Figure 4. Plots of bulk compositions of black phyllite and schist from the Smalls Falls Formation. (a), TOC vs total S. Regression lines for Jurassic Jet Rock Member of the Whitby Mudstone Formation and Holocene Black Sea sediments from Raiswell and Berner (1985) and Lyons and Berner (1992), respectively. (b), TOC vs Mo. Regression lines for anoxic and euxinic sediments of Black Sea, Framvaren Fjord (Norway), Cariaco Basin (offshore Venezuela), and Saanich Inlet (British Columbia) from Algeo and Lyons (2006) and references therein. Minimum Mo content for persistently euxinic (sulphidic) conditions from Scott and Lyons (2012). Note TOC in both plots is total non-carbonate carbon (organic C + graphitic C). (c), total S vs Mo. For samples from garnet zone, open symbols are from location 12; filled symbol is from location 13. Modern suboxic to euxinic marine sediments have total S <2.5 wt % and are not shown. Data for Bakken Formation from Scott et al. (2017b, Appendix).



positive correlations exist between Al-Sc and Ti-Sc ($R^2 = 0.90$ and 0.86, respectively), which suggest that Sc was relatively immobile assuming that Al and Ti were immobile. The variation in Th/Sc ratios of samples from location 12 may partly reflect Sc mobility related to channelized fluid flow during metamorphism, as discussed below.

A ternary plot of Th-Sc-Zr/10 (Fig. 7) shows that data for all samples from greenschist-grade outcrops have bulk compositions that fall in the field for continental arcs (Bhatia and Crook 1986). A source from a predominantly felsic arc is thus proposed, an interpretation supported by elevated Sn contents of several analyzed samples, which range up to 29 ppm (Table 2). For comparison (Fig. 8), median Sn in black shale globally is 3.9 ± 0.3 ppm for all host lithologies and 6.6 ± 0.4 ppm for terrigenous + volcanic-sedimentary lithologies (Ketris and Yudovich 2009); Sn in average shale is 3 ppm (Li and Schoonmaker, 2003). Tin concentrations above ca. 5 ppm in our Smalls Falls samples are unlikely to reflect significant mass loss during metamorphism, because these values are all in biotite-grade samples and because the greatest mass loss in pelitic schists tends to occur at higher grades, in upper greenschist- and amphibolite-facies rocks (e.g., Ague 1994a). The locally high Sn contents reported here, together with elevated Th/Sc ratios, suggest an evolved felsic plutonic and/or volcanic source within a former continental arc. In an earlier study, Cullers et al. (1997) reached a similar conclusion for the Smalls Falls Formation based on trace element ratios and REE patterns, albeit for a small data set of only two samples.

Assuming that the early Ludlow graptolite age for the Smalls Falls Formation is valid, a Middle Silurian or older arc is required as a source terrane. Potential candidates include Early Silurian rhyolite in the upper part of the Ascot Complex in southeastern Québec that has a U-Pb zircon date of 441 +7/-12 Ma (David and Marquis 1994). Rhyolites of the Stokes domain, within the Ascot Complex, are derived from an enriched magma with a significant continental crustal component (Tremblay et al. 1989); such evolved felsic magmas typically contain elevated Sn and may have associated tin ores (e.g., Cérný et al. 2005), although no Sn analyses are available for these rhyolites. Additional possible sources are older felsic-dominated arcs as represented by the Early Silurian Attean pluton in northwestern Maine (443 \pm 3 Ma; Gerbi et al. 2006) and by Middle Ordovician rhyolites of the Ascot Complex in southeastern Québec (460 \pm 3) Ma; David and Marquis 1994), but both of these sources are questionable due to the paucity of Paleozoic detrital zircon age peaks older than 437 Ma reported for the Smalls Falls Formation by Bradley and O'Sullivan (2006).

Deposition and diagenesis

In addition to provenance, the geochemical signature of organic-rich sediments is influenced by depositional parameters including sedimentary sorting, chemical and biological contributions, and the redox state of bottom waters and pore fluids (e.g., Arthur and Sageman 1994; Sageman and Lyons 2003). Given the moderate to pronounced metamorphic imprint on the Smalls Falls samples used in this study, and the lack of unmetamorphosed equivalents for analysis, it is not possible to evaluate all effects related to sedimentation. However, the mostly very low TOC yet locally high total S contents (Fig. 4a; Appendix) warrant discussion of possible controls including premetamorphic redox state(s).

Numerous geochemical proxies have been employed for evaluating depositional redox states (e.g., Jones and Manning 1994; Rimmer 2004; Tribovillard et al. 2006). However, most trace element proxies used in the past such as Ni/ Co, V+Cr, and V/(V+Ni) typically yield ambiguous or misleading results (Algeo and Liu 2020; Bennett and Canfield 2020), and hence are not adopted in this study. Among various proxies, one of the more robust for unmetamorphosed black shales is Mo content (Scott and Lyons 2012), which for all 21 samples from the Smalls Falls Formation ranges from less than the detection limit (2 ppm) up to 62 ppm. These relatively Mo-poor values suggest that the bottom waters during sedimentation were mainly suboxic to anoxic and lacked persistently euxinic (sulphidic) conditions, based on Mo concentrations of <100 ppm. It is possible that samples having less than 25 ppm Mo record oxic bottom waters (Scott and Lyons 2012), but such values could also reflect dilution by quartz or other clastic components (e.g., Sageman and Lyons 2003). Equally likely, Mo may have been partially lost from the sediments during late diagenesis and/or metamorphism, based on the studies by Wang et al. (2011) and Ardakani et al. (2016) who found widespread mobility of Mo associated with these processes. Also relevant to the paleoredox issue are the small pyrite framboids less than 8 µm in diameter (Fig. 3) that occur in one chlorite-zone sample from location 10 (Fig. 1), which are interpreted as primary grains that are preserved through metamorphism. The small size of these framboids suggests that euxinic bottom waters existed during sedimentation there, based on analogy with the size of framboidal pyrite grains in the euxinic water column and sediments of the modern Black Sea (Wilkin et al. 1997) and in unmetamorphosed organic-rich mudrocks that are attributed to euxinic depositional conditions (Wignall and Newton 1998). Uniformly low MnO contents (<0.01 wt %) further suggest that the Smalls Falls basin(s) had at least locally anoxic bottom waters (e.g., Quinby-Hunt and Wilde 1994). Smalls Falls samples from the chlorite and biotite zones have log ratios for U/Al and V/Al of 0.8 to 2.0 and 12 to 23, respectively, which if largely preserved through greenschist-grade metamorphism, imply deposition under a perennial oxygen minimum zone (see Bennett and Canfield 2020).

Additional insights into depositional conditions come from a plot of TOC vs Mo (Fig. 4b). Except for the Black Sea regression line, those shown for the other modern anoxic to euxinic basins have higher TOC contents and much higher Mo/TOC ratios. The Black Sea trend reflects a special case in which the basin is largely restricted from exchange with Mobearing oxic seawater, thus producing Mo-poor euxinic sediments. Different trends shown for the three other examples

Table 2. Basic statistics for trace elements and rare earth elements in the Smalls FallsFormation and in marine shale and black shale.

Elt	Median (range)	Avg marine	Median black	Median enrich-	Percent
Element	Smalls Falls Fm.	shale ¹	shale ²	ment factor ³	change
Li (ppm)	40.0 (30-100)	66	44 ± 2	0.87	-13
Be	<5	3	2.4 ± 0.2	no data	no data
Ga	19.0 (10-30)	19	20 ± 1	0.909	-9.1
Ge	2.0 (<2-2)	1.6	2.8 ± 0.2	0.683	-31.7
Sc	14.0 (6-21)	13	14 ± 1	0.957	-4.3
V	112.0 (61-217)	130	200 ± 10	0.536	-46.4
Cr	70.0 (30-100)	90	100 ± 7	0.7	-30
Zr	177.0 (97.3-331)	160	150 ± 10	1.129	12.9
Hf	5.0 (3-10)	5	3.8 ± 0.5	1.259	25.9
Nb	12.0 (7-18)	11	12 ± 1	0.957	-4.3
Та	0.80 (<0.5-1.3)	1.3	0.83 ± 0.09	not reported ⁴	not reported ⁴
Rb	137.0 (54.3-221)	140	93 ± 9	1.41	41
Sr	177.0 (47.0-330)	170	200 ± 10	0.847	-15.3
Cs	7.10 (2.1-13.6)	5	5.9 ± 1.0	1.151	15.1
Ba	436 (187-801)	580	560 ± 60	0.745	-25.5
Y	25.5 (13.8-45.6)	26	29 ± 1	0.841	-15.9
La	33.0 (11.7-48.9)	43	31 ± 2	1.019	1.9
Ce	71.2 (25.7-107)	82	61 ± 6	1.117	11.7
Pr	8.57 (3.20-12.3)	9.8	4.1 ± 0.5	not reported ⁴	not reported ⁴
Nd	31.3 (11.9-45.2)	33	25 ± 3	1.198	19.8
Sm	6.10 (2.6-8.8)	6.2	4.5 ± 0.5	1.297	29.7
Eu	1.32 (0.67-2.12)	1.2	1.1 ± 0.06	1.148	14.8
Gd	5.32 (2.72-8.82)	5.1	5.5 ± 0.5	0.926	-7.4
Tb	0.88 (0.48-1.27)	0.84	0.6 ± 0.02	not reported ⁴	not reported ⁴
Dy	4.91 (2.58-7.99)	4.7	3.9 ± 0.4	not reported ⁴	not reported ⁴
Ho	0.96 (0.47-1.58)	1.1	0.57 ± 0.04	not reported ⁴	not reported ⁴
Er	2.91 (1.54-5.17)	4.1	2.0 ± 0.1	not reported ⁴	not reported ⁴
Tm	0.49 (0.25-0.85)	0.44	0.30 ± 0.03	not reported ⁴	not reported ⁴
Yb	2.80 (1.5-5.3)	2.8	2.9 ± 0.2	0.924	-7.6
Lu	0.48 (0.23-0.83)	0.42	0.40 ± 0.01	1.148	14.8
Cu	41.0 (24-64)	45	100 ± 8	0.392	-60.8
Ni	58.0 (34-82)	50	84 ± 6	0.661	-33.9
Co	22.4 (12.6-38.6)	19	17 ± 2	1.261	26.1
Cd	0.30 (<0.2-1.3)	0.3	5.3 ± 1.3	0.054	-94.6
Zn	70.0 (6-132)	95	140 ± 20	0.478	-52.2
Pb	18.0 (11-34)	20	29 ± 2	0.594	-40.6
Mo	20.0 (<2-61)	2.6	18 ± 3	1.063	6.3
Ag	<1	0.07	1.9 ± 0.3	no data	no data
Au (ppb)	5.0 (<3-38)	2.5	8.8 ± 7.7	0.544	-45.6
Pd (ppb)	3.0 (<3-9)	1	no data	no data	no data
Pt (ppb)	5.0 (2-8)	1	no data	no data	no data

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Element	Median (range) Smalls Falls Fm.	Avg marine shale ¹	Median black shale ²	Median enrich- ment factor ³	Percent change
In	<0.2	0.1	4.6 ± 3.2	no data	no data
Bi	0.40 (0.1-2.5)	0.43	1.1 ± 0.8	0.348	-65.2
Tl	1.00 (0.6-1.5)	0.7	2.6 ± 1.1	0.368	-63.2
Sb	1.10 (<0.1-2.6)	1.5	3.6 ± 0.4	0.292	-70.8
As	15.0 (30-280)	13	27 ± 3	0.532	-46.8
Sn	2.0 (<1-29)	3	6.6 ± 0.4	0.29	-71
W	2.0 (1-4)	1.8	2.4 ± 1.9	not reported ⁴	not reported ⁴
Th	10.8 (4.20-17.4)	12	7.4 ± 0.6	1.396	39.6
U	6.49 (2.87-15.2)	2.7	14 ± 3	0.444	-55.6

Table 2.	Continued.
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¹Data for average marine shale from Li and Schoonmaker (2003).

²Data for median black shale (terrigenous and volcanic-sedimentary category) from Ketris and Yudovitch (2009).

³Enrichment factors are calculated by normalization to median Ti for both Smalls Falls Formation (4076 ppm) and global black shale (3900 ppm; Ketris and Yudovich 2009).

⁴Results not reported because number of analyses in Ketris and Yudovitch (2009) database are relatively limited (n < 700).

Note: For calculation purposes analyses below detection limit are assigned one half that value (Sanford *et al.* 1993).

(Saanich Inlet, Cariaco Basin, Framvaren Fjord) also record varying degrees of basin restriction and seawater ventilation of deep waters (Algeo and Lyons 2006). Although the slope of the trend for the Smalls Falls data is like that for Saanich Inlet, this similarity does not require a link between them in terms of depositional redox conditions.

Organic C and total S contents may change greatly during diagenesis. In this context, diagenesis includes both early biogeochemical cycling in sediments and later processes prior to, during, and after lithification. Raiswell and Berner (1987) found that TOC/S ratios in marine sediments progressively decrease with increasing vitrinite reflectance, yielding an average TOC/S ratio of approximately 1.9 and a ca. 30% loss of organic C at the beginning of thermal maturity and hydrocarbon generation, to an average TOC/S ratio of approximately 0.6 and total organic C loss of ca. 70% at the anthracite grade of organic metamorphism. These ratios reflect not only pyrite formation but also the loss of organic C by bacterial methane generation and hydrocarbon expulsion (Raiswell and Berner 1986). Ratios of TOC/S for the Smalls Falls Formation, for all metamorphic zones, average 0.41 ± 0.16 (Appendix). For comparison, Berner and Raiswell (1983) proposed a much higher organic C/pyrite S ratio of ca. 6.3 for sedimentary rocks of similar Late Silurian age. Based on these comparisons, together with TOC and sulphur data for modern and ancient anoxic to euxinic sediments (Fig. 4a), we estimate that Smalls Falls strata lost approximately 2 to 6 wt % organic C during diagenesis.

Diagenetic effects on trace elements in organic-rich mudstones and black shales have received considerable attention. Hannigan and Basu (1998) found depletions in Th and LREE, and enrichment in U during late diagenesis, based on data for immature (20°-50°C), mature and oil-bearing (50°-140°C), and post-mature (>200°C) strata of Late Ordovician age from Québec, Ontario, and New York, respectively. Abanda and Hannigan (2006) concluded that V, Cr, Ba, Th, Zr, and REE were depleted in thermally mature and post-mature samples, relative to concentrations in immature samples, and that the post-mature samples have higher U, Sr, and Co by comparison. In contrast, Scott et al. (2017a) reported a loss of ca. 30% organic C in one sample from the Late Devonian-Early Mississippian Bakken Formation in North Dakota, attributed to thermal effects during late diagenesis at or above the thermal window for oil generation (~60-120°C), but no significant variation in trace metal/ TOC ratios for other samples having burial conditions ranging from thermal immaturity to the peak oil window. However, significant amounts of V and Ni, and locally other metals such as Mo, Cu, and Zn, may be lost at thermal maturity based on their high concentrations in some crude oils (Lewan 1984; Ventura et al. 2015; Yang et al. 2018b). Given the pervasive metamorphic overprint on the Smalls Falls samples, a proper evaluation of possible diagenetic loss of selected metals such as Cr, V, and Zn cannot be done. However, loss of Mo is possible given that chlorite-zone sample JS-13-10B contains 28 ppm (Appendix), which is slightly low based on the presence in this sample of small pyrite framboids (Fig. 3) that suggest at least intermittent euxinic conditions and a related Mo concentration of 30 to 70 ppm in coeval sediments (Scott and Lyons 2012).



Figure 5. Shale-normalized plots of rare earth element abundances in black phyllite and schist from the Smalls Falls Formation. (a), Samples from chlorite and biotite zones. (b), Samples from garnet zone compared to those from chlorite + biotite zones. (c), Samples from staurolite-andalusite zone compared to those from chlorite + biotite zones. Normalization values are average Post-Archean Australian Shale (PAAS) from Taylor and Mc-Lennan (1985).

Some unmetamorphosed black shales show evidence for local redistribution of REE. For example, Lev et al. (1999) described altered La/Sm ratios and variable Ce/Ce* values beyond the range of those reported for unaltered sediments and sedimentary rocks, reflecting one or more periods of dissolution and reprecipitation of REE-bearing minerals (e.g., monazite, apatite) at the sediment-water interface and/ or during early to late diagenesis (see Lev and Filer 2004, and references therein). However, all Smalls Falls samples analyzed in our study have Ce/Ce* values of 0.95 to 1.01 (Appendix), which are typical for unaltered sedimentary rocks (e.g., Taylor and McLennan 1985), thus arguing against appreciable Ce redistribution during diagenesis (or metamorphism). Moreover, five samples from the chlorite zone lack positive Eu anomalies (Eu/Eu* = 0.90-0.97; Appendix), suggesting little if any influence on precursor sediments by reduced hydrothermal fluids during diagenesis. Minor enrichment of MREE observed in samples from the chlorite and biotite zones, relative to PAAS (Fig. 5a), may reflect mobility of these elements during low-grade metamorphism and/or earlier diagenesis (e.g., Lev and Filer 2004).

Metamorphism

Geochemical trends

Varying degrees of sedimentary sorting and primary detrital vs biological contributions, plus the limited database of our study, preclude a detailed quantitative assessment of metamorphic effects on major-element compositions of the Smalls Falls samples. However, some generalizations can be made. These are typically based on element normalization to either Al or Ti and on the assumption that these two elements are conservative components in the sample set, a hypothesis supported by low variance of the Al/Ti ratio among all samples analyzed regardless of metamorphic grade (avg = 17.8 ± 1.51).

Changes in average Si/Al ratios show a general decrease with increasing metamorphic grade (Appendix). These changes are, as follows: (1) chlorite zone = 5.73 ± 1.98 (*n* = 5); (2) biotite zone = 3.88 ± 0.90 (*n* = 7); (3) garnet zone = 3.91 (excluding location 12); and (4) staurolite-andalusite zone = 2.97 ± 0.82 (*n* = 5). This progressive decrease in Si/ Al ratios with increasing metamorphism is consistent with the results of more detailed studies in other metamorphic terranes that suggest silica is increasingly lost with increased grade of metamorphism (e.g., Ague 1994a, 2017; Ferry 1994; Masters and Ague 2005). The lowest Si/Al ratios found for samples from the staurolite-andalusite zone are thus attributed here to a major loss of silica that accompanied the P-T-X conditions of this metamorphism. Other major elements in the Smalls Falls database lack significant variations in Al-normalized ratios, except for anomalously high Fe/Al of 1.37 and 1.45 in two samples from the garnet zone at location 12, relative to average ratios of 0.51 to 0.87 for the other three zones (Appendix). All three samples from location 12 have the highest total S contents (4.54-9.67 wt %), reflected



Figure 6. Binary Zr/Sc vs Th/Sc plot of bulk compositions of black phyllite and schist from the Smalls Falls Formation. For samples from garnet zone, open symbols are from lo-cation 12; filled symbol is from location 13. Discriminant values from McLennan *et al.* (1993).

in relatively abundant pyrite (Fig. 2c).

Data for the Smalls Falls Formation also suggest an overall loss of TOC with increasing metamorphic grade. Average TOC/Al ratios for the chlorite, biotite, garnet, and staurolite-andalusite zones are 0.25 ± 0.05 , 0.10 ± 0.02 , $0.16 \pm$ 0.08, and 0.09 ± 0.02, respectively. The greatest TOC loss appears to have occurred as a result of P-T conditions that existed within the biotite zone, which in western Maine had pressures below 330 ± 40 MPa and temperatures below 525 \pm 30°C as determined for the staurolite-andalusite zone in this region by Holdaway et al. (1982). Average total S/Al ratios also decrease with increasing metamorphic grade in our Smalls Falls database, from 0.59 ± 0.24 , to 0.33 ± 0.11 , to 0.30 (location 13 only), to 0.28 ± 0.08 , in the chlorite, biotite, garnet, and staurolite-andalusite zones, respectively; these results indicate that the relatively high absolute total S contents among Smalls Falls samples-including up to 3.9 wt % in the chlorite zone-are unrelated to enrichment due to mass loss during metamorphism, which should produce the opposite trend. Nitrogen (as ammonium ion in micas) also shows a progressive loss in concentration with increasing metamorphic grade in samples from the Smalls Falls Formation (Van Baalen 2006).

Table 3 lists median enrichment factors and percent changes for selected trace elements and REE in the four metamorphic zones sampled in this study. These values were calculated on a Ti-normalized basis relative to the global median black shale of Ketris and Yudovich (2009); Al was not used because this element is not listed in their database. A statistical evaluation of the observed geochemical variations has been performed by the Tukey-Kramer analysis of variance method (Kramer 1956; Dunnett 1980),



Figure 7. Ternary Zr-Th-Sc plot of bulk compositions of black phyllite and schist from the Smalls Falls Formation. For samples from garnet zone, open symbols are from lo-cation 12; filled symbol is from location 13. Fields from Bhatia and Crook (1986).

using JMP software v. 14.3.0 (SAS Institute Inc. 2018). The Tukey-Kramer test analysis controls for group variance and differences in group sample size by calculating significance difference tests for each group of sample means and medians. Results of this analysis, for selected trace elements in our Smalls Falls database (V, Cr, Y, Cu, Ni, Zn, Pb, Th, U), verify the differences in Ti-normalized values reported here for the four metamorphic zones.

The Ti-normalized values show moderate to large decreases (-23.1 to -49.8%) for V, Cr, Cu, Ni, Zn, Pb, Sb, and U in the chlorite zone, attributed here to undefined factors during sedimentation and to selective mobility during diagenesis. Systematic small to extreme decreases (-14.5 to -99.0%) were found for Ba, Sb, Au, and U with increasing metamorphic grade (biotite through staurolite- andalusite), together with less-systematic moderate to large decreases (-35.4 to -61.1%) for V and As. Molybdenum displays progressive increases until the staurolite-andalusite zone in which values greatly decrease (-94.7%). Values for V, Cr, and U display uniformly moderate to large decreas-es throughout all four metamorphic zones. Enrichment factors and percent changes for Y and REEs show moderate to large decreases from the biotite to garnet zones (e.g., -47.3% for La), probably reflecting the breakdown of apatite and monazite; a similar decrease for Sr (-43.2%) may record metamorphic decarbonation reactions. Moderate to large increases calculated for Rb (+28.1 to +61.5%) and Th (+39.1 to +47.3%) from the biotite to staurolite-andalusite zones likely reflect alkali introduction and mass loss, respectively, during metamorphism (e.g., Ague 1994a).



Figure 8. Histogram of tin contents of black phyllite and schist from the Smalls Falls Formation. Shown for comparison are values for global median black shale compositions (Ketris and Yudovich 2009) for two lithologic categories of host rocks: (1) chert + carbonate + terrigenous + volcanic-sedimentary rocks (n = 8040), and (2) terrigenous + volcanic-sedimentary rocks (n = 4480).

Organic carbon and iron sulphide transformations

Detailed mineralogical studies have documented the residence of many trace elements in OM and iron sulphides. For example, Ripley et al. (1990) analyzed the unmetamorphosed Upper Devonian-Lower Mississippian New Albany Shale in Indiana and found that kerogen contains 20% of the Mo and 50% of the Ni, whereas iron sulphide contains ca. 56 to 85% of the Mo, 45% of the Ni, and essentially all of the Zn, Cu, and Cd; Pb resides in selenide and pyrite. Chappaz et al. (2014) reported that 80 to 100% of the Mo in euxinic shales resides in the non-sulphide matrix, whereas a maximum of 20% occurs in pyrite. A more recent study of Mo speciation in Ordovician black shales of southern Québec by Ardakani et al. (2016) documented the presence of $61 \pm 5\%$ Mo(VI) in OM and $39 \pm 5\%$ Mo(IV)S, as molybdenite. Abanda and Hannigan (2006) reported on leaching experiments of correlative black shales in Québec, Ontario, and New York, and related mass-balance calculations that suggest about 22% of the trace element and REE contents are contained in the organic fraction. Given the abundances of trace elements and REE in the OM of many unmetamorphosed black shales, the common transformation of largely amorphous OM into variably crystalline graphite during metamorphism may greatly alter the mineralogical residence and whole-rock budget of these elements. The OM to graphite transformation, which depends on several parameters including temperature, pressure, and metamorphic fluid composition (especially f_{CH4}), generally begins in the greenschist facies and is complete, i.e., only fully ordered graphite is present, within the upper amphibolite facies (e.g., Nishimura et al. 2000). However, in some terranes, graphitization is complete by the upper greenschist facies (Wang 1989), whereas in others no correlation exists between metamorphic grade and the degree of graphite crystallinity (Wintsch et al. 1981; Kříbek et al. 2008). Nonetheless, the well-documented graphitization of OM during the metamorphism of black shales elsewhere suggests that this process governed, at least in part, the uniformly negative median enrichment factors and percent changes observed for U, V, Cr, and Ni in our samples from the Smalls Falls Formation (Table 3).

Metals in black shales that are contained mainly in iron sulphide minerals can also be lost during the metamorphic recrystallization of pyrite and the transformation of pyrite to pyrrhotite. Recrystallization of pyrite commonly releas-es numerous trace elements, based on in situ laser ablationinductively coupled plasma-mass spectrometry (LA-ICP-MS) analysis of different pyrite generations in black shale (Large et al. 2007, 2011; Gregory et al. 2016). In pelitic schists, the pyrite to pyrrhotite transformation commonly begins in the middle to upper greenschist facies with the breakdown of chlorite (as a major source of fluid) and is complete by the upper greenschist to upper amphibolite facies, depending on various physicochemical parameters (e.g., Ferry 1981; Tomkins 2010). In our samples of metamorphosed black shale from the Smalls Falls Formation, pyrrhotite is absent from the chlorite zone, rare in the biotite zone, uncom-mon in the garnet zone, and predominant in the staurolite- and alusite zone (Fig. 2). The chalcophile elements Cu, Zn, Sb, and As, as well as Mo, Ni, and Co (siderophile), all show uniformly large decreases in median enrichment factors and percent changes in the staurolite-andalusite zone, relative to the garnet zone; Au and Sb values progressively decrease above the chlorite zone (Table 3). Considered overall, these decreases suggest preferential loss of chalcophile and some siderophile elements to the fluid phase, related to the meta-morphic recrystallization of pyrite, and to the metamorphic transformation of pyrite to pyrrhotite, with limited ability of the latter mineral to incorporate appreciable amounts of these trace elements. This interpretation is consistent with findings by other workers, based on in situ geochemical data obtained via LA-ICP-MS. For example, Large et al. (2011) described major depletion of Pb, Cu, Zn, Sb, Ag, Mo, Bi, Te, and Tl in carbonaceous shales during the lowgrade metamorphic recrystallization of pyrite, and substantial loss of Au, As, S, Sb, Te, Cu, Zn, Mo, Bi, Tl, and Pb in similar rocks at higher grades during the pyrite to pyrrhotite transition. Pitcairn et al. (2010a, 2015) reported pervasive loss of As, Au, Hg, and Sb in amphibolite-facies rocks of the Otago and Alpine schists (New Zealand) and the Dalradian Supergroup (Scotland) that contain only pyrrhotite, relative

	Chlorite	(n - 5)	Biotite	(n - 7)	Garnet	(n-4)	Staur	olite-
	Cilionia	(n - 3)	Diotite	(n-7)	Garriet	(n - 4)	<u>Andalusi</u>	te (<i>n</i> = 5)
Element	Median	Percent	Median	Percent	Median	Percent	Median	Percent
Element	EF ³	Change						
V	0.5023	-49.8	0.6003	-40.0	0.5054	-49.5	0.4593	-54.1
Cr	0.5707	-42.9	0.6073	-39.3	0.7490	-25.1	0.6164	-38.4
Rb	1.2765	27.7	1.2810	28.1	1.6146	61.5	1.5507	55.1
Sr	1.0302	3.0	1.0961	9.6	0.5677	-43.2	0.5177	-48.2
Ba	0.8887	-11.1	0.8551	-14.5	0.6961	-30.4	0.6169	-38.3
Y	0.9231	-7.7	0.9227	-7.7	0.7610	-23.9	0.7624	-23.8
La	1.1574	15.7	1.1072	10.7	0.5273	-47.3	0.8417	-15.8
Eu	1.1382	13.8	1.2996	30.0	0.8862	-11.4	0.8395	-16.0
Yb	0.9722	-2.8	0.9573	-4.3	0.8414	-15.9	0.8444	-15.6
Cu	0.6135	-38.6	0.3482	-65.2	0.5809	-41.9	0.2679	-73.2
Ni	0.7474	-25.3	0.6584	-34.2	0.9394	-6.1	0.4240	-57.6
Co	1.2990	29.9	0.8899	-11.0	2.0625	106.0	0.9629	-3.7
Zn	0.6507	-34.9	0.3233	-67.7	0.6547	-34.5	0.3258	-67.4
Pb	0.7692	-23.1	0.5048	-49.5	0.5421	-45.8	0.6273	-37.3
Mo	1.2683	26.8	1.8328	83.3	2.5021	150.0	0.0532	-94.7
Au	1.1091	10.9	0.6248	-37.5	0.1930	-80.7	0.1193	-88.1
Sb	0.7293	-27.1	0.4067	-59.3	0.0857	-91.4	0.0097	-99.0
As	2.7541	175.0	0.5091	-49.1	0.6455	-35.4	0.3887	-61.1
Sn	0.2987	-70.1	0.9858	-1.4	0.0858	-91.4	0.2900	-71.0
Th	1.4894	48.9	1.4728	47.3	1.4366	43.7	1.3914	39.1
U	0.5255	-47.4	0.7610	-23.9	0.4829	-51.7	0.2125	-78.7

Table 3. Median trace element and REE element enrichment factors and percent changes in Small Falls Formation at different metamorphic grades¹.

¹Data are normalized to values for median black shale (terrigenous + volcanic-sedimentary category) of Ketris and Yudovitch (2009).

²TOC values are total non-carbonate carbon that includes both organic C (analyzed) and graphitic C (calculated).

³Enrichment factors are calculated by normalization to Ti contents (ppm) in both Smalls Falls Formation and median black shale (Ketris and Yudovitch 2009).

Note: For calculation purposes analyses below detection limits are assigned one half that value (Sanford et al. 1993).

to pyrite-bearing greenschist-facies equivalents in both areas. Following these studies, we suggest that metamorphic devolatilization of Smalls Falls strata transferred numerous trace elements to the fluid phase that were partly to largely removed from the local environment.

The presence of positive Eu anomalies in some of the Smalls Falls samples is consistent with local influence by reduced metamorphic fluids (e.g., Bau 1991). Importantly, all chlorite-zone samples lack positive anomalies defined as Eu/ $Eu^* \ge 1.10$ (Appendix), thus ruling out redox influence on Eu²⁺/Eu³⁺ equilibria by anoxia or euxinia in synsedimentary to early diagenetic pore fluids (see Slack et al. 2015). Positive Eu anomalies are rare in metasedimentary rocks that are spatially unrelated to hydrothermal ore deposits, thus suggesting a link to the mineralogy and composition of the Smalls Falls strata. Occurrence of abundant iron sulphide within rocks of this formation, together with minor OM

and/or graphite, implies that metamorphic fluids in equilibrium with these rocks were reducing and had temperatures greater than 250°C, and locally produced Eu²⁺-dominant speciation and resulting positive Eu anomalies.

Channelized fluid flow

In addition to mineralogical and geochemical effects from organic carbon and iron sulphide transformations in the Smalls Falls Formation, evidence for channelized fluid flow exists in one outcrop. Samples from location 12, north of the town of Bingham (Fig. 1; Table 1), are atypical in having anomalously high Fe/Al and low La/Yb ratios, compared to the one sample from location 13 in the same garnet zone and to those analyzed from the other metamorphic zones (Fig. 9). The three samples from location 12 are also compositionally distinct in having the highest total S contents



Figure 9. Binary Fe/Al vs La/Yb plot of bulk compositions of black phyllite and schist from the Smalls Falls Formation.

(4.54-9.67 wt %; Fig. 4) and highest As (280 ppm) and Co (38.6 ppm), as well as small positive Eu anomalies (Eu/Eu* = 1.10-1.45). These data suggest that enrichment in pyrite, as a host for the As and Co, and deposition of this pyrite, was from reduced fluids. Predominance of foliated pyrite in these samples (Fig. 2c) indicates that these fluids were introduced before penetrative deformation ended. The distinctly low La/Yb ratios imply preferential loss of LREE from the pyritic schist at this site. The sample with the highest Eu/Eu* value of 1.45, from location 12, also has the highest Na₂O content (1.82 wt %; Appendix), suggesting a metamorphic influx of Na and related growth of sodic plagioclase that sequestered Eu²⁺ and a resulting positive Eu anomaly (see Ague 2017). However, compared to the other metamorphic zones, these three samples lack elevated Si/Al ratios (Appendix), consistent with an absence of introduced silica (as quartz) although limited sampling and surface weathering of this outcrop make it difficult to rule out unrecognized metamorphic quartz veins at or near the sample sites.

The geochemical data reported here for location 12 are interpreted to reflect channelized fluid flow during metamorphism. This process has been studied by many workers, because of its importance in driving metamorphic reactions and mass changes in the crust (e.g., Ferry 1994; Ague 1994b, 2017; Skelton *et al.* 1995; Pitcairn *et al.* 2010b). Our reconnaissance data for location 12 are consistent with the results of these other, more detailed, studies including for REE (cf. Ague 2017). A thorough evaluation of channelized fluid flow in this outcrop, including possible lithologic and/ or structural controls, awaits further sampling and analysis.

Regional considerations

On the northwestern margin of the CMB, the Smalls Falls Formation has conformable contacts with both the underlying Perry Mountain Formation and overlying Madrid Formation. Point counts of metasedimentary rocks from these formations by Moench (2006) indicate high sediment maturity within the Perry Mountain Formation and the Smalls Falls followed by a marked decrease in maturity in the overlying Madrid Formation. These data suggest that the onset of anoxia reflects changes in the water column rather than in local geologic variables (e.g., provenance). In contrast, the Smalls Falls–Madrid transition constitutes the more significant local geologic change, which has been interpreted to record the switch from a northwestern source to a southeastern one (Bradley *et al.* 2000).

Regional considerations point to the emergence of a landmass southeast of Central Maine during the Silurian (e.g., Bradley *et al.* 2000). Specifically, a thick section of sandstones—the Mayflower Hill Formation (of the Vassalboro Group; Marvinney *et al.* 2010; Ludman *et al.* 2020)—conformably overlies the dominantly pelitic Wenlock Waterville Formation. Together, these formations define a coarseningupwards sequence prior to termination of sedimentation in the Late Silurian. Plutons of Ludlovian age farther southeast mark closure of the Fredericton Trough on the Dog Bay Line, the presumed mechanism for Silurian uplift and emergence (Reusch and van Staal 2012).

Reusch and van Staal (2012) offered specific mechanisms for how emergence of the southeastern source-the Brunswick subduction complex-might trigger anoxia on the northwest margin of the CMB. These mechanisms include a bathymetric high that cut off exchange with the open ocean, a fresh-water lid that impeded lateral and vertical circulation, and nutrient loading that enhanced productivity and eutrophication. Basin contraction brought the southeastern landmass closer to the site of Smalls Falls sedimentation, until black shale deposition was replaced by a mix of southeasterly derived siliciclastic and carbonate turbidites within the Madrid Formation (Moench 2006). Widespread algal blooms, analogous to those described for other Paleozoic black shales (e.g., Grimm et al. 1997; Macquaker et al. 2010), may have provided the OM responsible for the reducing conditions in bottom waters and pore fluids inferred here for the Smalls Falls sediments.

Potential modern analogs

Given the geochemical constraints pointing to a variably anoxic depositional environment for the Smalls Falls Formation, several potential modern analogues can be considered. First, upwelling zones such as those offshore Peru and Namibia are rejected as candidates, because the reduced sediments there are locally highly phosphatic and include phosphorites with P_2O_5 contents that exceed 18 wt % (e.g., Föllmi 1996), which contrast with the less than 0.2 wt % P_2O_5 present in Smalls Falls strata (Appendix). Second, very local restricted basins, such as Framvaren Fjord in Norway that has extremely high dissolved sulphide (H_2S) in the water column (e.g., Skei 1983), are far too small to serve as valid analogs.

Other modern suboxic to euxinic basins to evaluate including the Cariaco Basin, Orca Basin, Southern California Borderland, and Black Sea (cf. Algeo and Tribovillard 2009). However, available geochemical data for the first three of these basins indicate that their contained sediments have uniformly low concentrations of total S or sulphide S (<2.5 wt %; Leslie et al. 1990; Hurtgen et al. 1999; Lyons et al. 2003; Figueroa et al. 2018). These low values contrast with our data for Smalls Falls strata (excluding pyrite-rich samples from location 12) that show ~2 to 4 wt % total S including up to 3.9 wt % total S in chlorite-grade samples that is unrelated to mass loss during metamorphism. However, the Black Sea differs in having older Holocene sediments (sapropels) with up to 8.3 wt % total S (Hirst 1974), although younger (<3200 yr) Holocene sediments there contain less than 2.5 wt % sulphide S (Lyons and Berner 1992). The Black Sea has a history of extension-related subsidence and a current foreland tectonic setting, like the ancestral CMB, and has been cut off from oxygenated Mediterranean waters throughout its ~1200-km width (Lyons and Berner 1992). This tectonic setting and local presence of sulphide-rich sediments in the Black Sea suggests that it may serve as a potential analog of the basin(s) in which strata of the Smalls Falls Formation accumulated. If this comparison is valid, it could also explain the anomalously low Mo contents of the Smalls Falls samples, relative to those of the Black Sea (Fig. 4b), by limited ventilation of Mo-bearing oxygenated seawater into the CMB during the Ludlovian.

Suggestions for future work

This reconnaissance study involved the analysis of only 21 samples from one stratigraphic unit (and correlatives) ca. 300 km in strike length (Fig. 1). As a result, only broad interpretations can be made, given the likelihood of unrecognized differences in trace element and REE concentrations at single locations and even within thin (<5 cm) beds, as a function of sedimentary sorting, rapid variations in redox state of bottom waters, and superimposed effects of diagenesis and metamorphism. Future work on the Smalls Falls Formation should obtain analyses of many more samples in a systematic approach that tests these parameters, including our hypothesis for synmetamorphic channelized fluid flow proposed at location 12 in western Maine. Worthwhile also would be in situ analysis, via LA-ICP-MS, of trace elements in pyrite and pyrrhotite (and other sulphides) using samples from different metamorphic grades, as a means to better evaluate the loss of metals during the transformation of pyrite to pyrrhotite.

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Any use in this paper of trade, firm, or product names is for descriptive purposes only and does not imply endorsement by the U.S. Government.

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APPENDIX

Appendix Tables A1 and A2 on following pages.

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ted ratios fo
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jor element
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-rock analy
1. Whole
Table A

TOC/ Ti		5.40	2.45	4.29	5.03	4.66	4.37	1.15		1./0	2.25	1.17	1.92	1.83	1.85	1.72	1.79	0.32	2.08	3.79	1.21	4.56	2.91	1.54	טר ר ר	1 8 1	1.35	1111	1.74	1.66	0.45	
TOC/ Al		0:30	0.18	0.24	0.27	0.25	0.25	0.05		01.0	0.13	0.08	0.11	0.10	0.11	0.10	0.10	0.02	0.11	0.24	0.07	0.22	0.16	0.08	C F 0	71.0	0.08	0.06	0.10	0.09	0.02	
Fe/Al		0.86	1.00	0.91	06.0	0.69	0.87	0.11		0.72	0.64	0.64	0.21	0.47	0.61	0.56	0.55	0.17	1.45	1.37	0.91	0.75	1.12	0.34	0.20	07.0	0.42	0.49	0.54	0.51	0.13	4
Si/Al		4.56	6.01	8.89	5.50	3.67	5.73	1.98	201	4.00	3.89	5.70	4.01	3.03	3.25	3.22	3.88	06.0	4.59	4.13	4.74	3.91	4.34	0.39	L V V	717.F	2.74	2.47	2.79	2.97	0.82	
TOC/S		0.53	0.23	0.28	0.60	0.95	0.52	0.29		17.0	0.36	0.26	0.67	0.36	0.24	0.35	0.35	0.15	0.12	0.15	0.09	0.73	0.27	0.30		00.0	0.34	0.24	0.30	0.34	0.13	
CG		0.68	0.08	0.27	0.70	1.05	0.56	0.38	10 0	10.0	0.02	0.01	0.00	0.00	0.00	0.00	0.01	0.01	0.00	0.00	0.00	1.01	0.25	0.51	0.02	0.0	0.01	0.01	0.01	0.01	0.01	- 1- 1
TOC		1.85	0.88	06.0	1.48	1.84	1.39	0.48		c/.U	1.01	0.46	0.82	0.88	0.89	0.88	0.81	0.18	0.70	1.48	0.43	1.53	1.03	0.55	0.03		0.79	0.63	0.97	0.86	0.15	۔ 9ر
Co		1.17	0.80	0.63	0.78	0.79	0.83	0.20		0.72	0.99	0.45	0.82	0.88	0.89	0.88	0.80	0.18	0.70	1.48	0.43	0.52	0.78	0.48	000	00.0	0.78	0.62	0.96	0.85	0.15	
CO ₂		0.93	0.11	0.37	0.96	1.45	0.76	0.53		70.0	0.03	0.01	<0.003	<0.003	<0.003	<0.003			<0.003	<0.003	<0.003	1.39				100	0.01	0.02	0.01	0.02	0.01	
C^{I}		2.10	0.91	1.00	1.74	2.24	1.60	0.62	t	0./4	1.02	0.46	0.82	0.88	0.89	0.88	0.81	0.18	0.70	1.48	0.43	1.91	1.13	0.68	100		0.79	0.64	0.97	0.87	0.15	1.4.4
S^{T}		3.51	3.88	3.22	2.48	1.94	3.01	0.79	ç	24.0	2.80	1.75	1.22	2.46	3.69	2.48	2.55	0.87	5.79	9.67	4.54	2.11	5.53	3.16	77 1	113	2.29	2.68	3.26	2.80	0.94	L COL
Total		97.30	97.59	97.09	97.30	97.21	97.30	0.18	100 10	100.49	100.78	100.76	99.70	99.66	99.49	96.64	99.65	1.43	98.76	99.07	99.91	96.52	98.57	1.45	00 00	111101	60.66	99.21	99.12	99.59	0.86	
IOI		9.39	9.97	6.58	6.38	7.49	7.96	1.64		06.0	5.24	4.45	5.15	7.04	8.12	n.a.	5.99	1.36	6.65	12.4	5.28	4.80	7.28	3.50	000	1.4.U	4.11	4.68	5.32	4.67	1.04	
TiO_2	%	0.57	0.60	0.35	0.49	0.66	0.53	0.12		60.0	0.75	0.65	0.71	0.80	0.80	0.85	0.75	0.07	0.56	0.65	0.59	0.56	0.59	0.04	070	0.00	10.07	0.95	0.93	0.89	0.12	00
P_2O_5	wt	0.09	0.11	0.08	0.11	0.19	0.12	0.04		11.0	0.11	0.15	0.09	0.11	0.11	0.07	0.11	0.02	0.09	0.03	0.10	0.10	0.08	0.03		70.0	0.05	0.05	0.05	0.05	0.02	I
K_2O		2.81	2.10	1.41	2.73	3.40	2.49	0.76		76.7	3.47	2.29	3.10	4.21	4.09	4.34	3.49	0.77	3.69	3.99	2.06	2.68	3.11	0.89	10 0	2 10	4.07	3.98	4.12	3.64	0.58	L
Na ₂ O		1.22	1.28	1.40	0.47	1.56	1.19	0.42	, , ,	C1.1	1.12	0.98	1.33	1.09	0.77	n.a.	1.07	0.19	0.43	0.69	1.82	0.69	0.91	0.62	10.0	1 10	1.04	0.97	0.97	1.07	0.23	
CaO		2.15	2.13	2.11	2.69	3.01	2.42	0.41		CU.2	1.94	2.19	2.39	2.16	1.26	1.40	1.91	0.42	0.43	0.66	2.57	3.75	1.85	1.59	100	150	0.099	1.12	0.53	1.00	0.35	- [-4-4
MgO		2.61	1.08	1.09	2.82	3.10	2.14	0.98		C0.7	2.72	3.53	3.18	3.26	2.58	2.06	2.85	0.50	2.29	2.13	3.40	3.27	2.77	0.65	0.02	2 1 2	1.67	3.24	1.25	2.04	1.07	L C T
MnO		0.06	0.02	0.02	0.06	0.05	0.04	0.02		0.04	0.04	0.06	0.05	0.06	0.06	0.05	0.05	0.01	0.02	0.02	0.08	0.31	0.11	0.14		70.0	0.04	0.07	0.03	0.05	0.02	141111
$\mathrm{Fe_2O_3}^\mathrm{T}$		7.50	7.12	4.90	7.05	7.15	6.74	1.05	L L	+C./	6.99	5.26	2.30	6.03	7.30	7.38	6.11	1.87	12.7	12.1	8.21	7.56	10.14	2.63	1 12	201 517	6.15	7.55	7.72	7.23	2.37	1
Al_2O_3		11.5	9.38	7.15	10.3	13.7	10.41	2.43		6.01	14.5	10.9	14.7	16.9	15.9	17.3	14.87	2.16	11.6	11.7	11.9	13.4	12.15	0.84	C 7 I	201	19.5	20.2	18.8	18.48	2.39	
SiO ₂		59.4	63.8	72.0	64.2	56.9	63.26	5.76		C.CO	63.9	70.3	66.7	58.0	58.5	63.2	63.44	4.32	60.3	54.7	63.9	59.4	59.58	3.79	15	7 V S	60.5	56.4	59.4	60.46	6.55	
Sample		JS-13-10A	JS-13-10B	JS-13-10C	JS-13-10D	JS-13-10E	Average	Stdev	V 1 2 1 JI	W11-61-6(JS-13-11B	JS-13-11C	JS-13-11D	JS-13-11E	JS-13-11F	JS-13-11G*	Average	Stdev	JS-13-12A	JS-13-12B	JS-13-12C	JS-13-13B	Average	Stdev	IC 13 144	IS 13 14B	IS-13-14C	IS-13-14D	JS-13-14E	Average	Stdev	Tota Tota

*Major-element analyses by ICP-AES; insufficient sample for XRF analysis.

Sample	Li	Ga	Ge	Sc	>	Cr	ŊŊ	Та	Zr	Ηf	Th	n	Ni	C	Cu	Pb	Zn	Cd 4	Au F	d Þ	t A:	s St	W Q	o Sr	M	E	Bi	Cs
									ıdd	n									P	þþ					bpm			
JS-13-10A	60	16	1	10	88	50	11	0.8	159	S.	9.6	9.29	55	22.9	51	28	92	<0.2	5	5	8	70 2	.3	20	2	1 0.4	3 0.3	4.0
JS-13-10B IS 12 10C	40	13		6 (81	70	1	0.8	295	6,	11.11 Č	6.49	42	15.9	57	18	53	<0.2	6 (4 (4 i	70 10	2 0	× 58	U .	1 0.	0.2	3.1
JS-13-10C	30	10	Γ	9	61	30	2	0.25	122	4	6.3	4.16	36	14.1	33	12	49	0.5	9	ŝ	n	40 1	×.	×	Ţ	1	0.7	2.1
JS-13-10D	50	15	1	6	76	40	10	0.6	167	S	8.3	5.54	59	16.4	51	22	72	0.3	38	Э	5	50 2		14	5	1 0.9	0.2	4.0
JS-13-10E	60	20	2	12	93	80	12	0.8	149	Ŋ	10.8	6.10	62	22.4	39	21	81	<0.2	S	З	9	50 2	2	16	7	1 1.0	0.2	4.7
Average	48.0	14.8	1.2	9.2	79.8	54.0	10.2	0.65	178.4	5.6	9.22	6.32	50.8 1	8.34 4	46.2 ź	20.2 6	9.4	1	3.0	3.6 5	6 56	0 2.1	2 19.	2	1.0	0.80	0.22	3.58
Stdev	13.0	3.7	0.4	2.2	12.4	20.7	1.9	0.24	67.4	1.9	1.97	<i>I.88</i>	11.3	4.03	9.9	5.8 1	8.3	I	4.1	i 0.0	.5 13	.4 0.1	.5	4	0.0	0.16	0.04	1.00
JS-13-11A	30	19	2	15	151	70	12	0.7	193	ŝ	10.8	12.1	63	21.7	40	16	48	0.4	9	6	8	50 2	9.	44	5	3 0.5	3 0.6	7.5
JS-13-11B	30	18	7	14	122	70	13	0.8	172	5	11.3	15.2	69	26.2	53	27	132	1.3	~	4	4	30 2	0	26	3	1 0.5	9.0.6	7.1
JS-13-11C	30	15	2	10	76	50	12	0.7	331	10	10.8	7.42	42	12.6	34	12	74	0.5	S	\$	4	30 1		13	4	2 0.5	3 0.7	7.8
JS-13-11D	30	23	2	14	131	60	15	0.8	251	~	12.1	8.54	34	13.0	38	13	28	0.4	9	4	°~ 9	30 1	-	36	81	2 0.9) 1.4	8.3
JS-13-11E	40	25	2	15	136	70	15	0.9	228	~	13.8	11.8	68	18.6	41	18	72	0.3	9	4	°~ 9	30 1	3	50	62	2 1.0	0 1.8	8.2
JS-13-11F	40	25	7	17	190	80	15	0.8	202	9	13.4	13.1	71	23.8	53	21	45	0.5	×	4	5	30 1	×.	50	8	3 1.	2.5	7.1
JS-13-11G	40	27	7	19	217	90	17	1.0	206	9	14.6	14.2	58	16.2	39	18	53	0.4	9	3	~ ~	30 2		51	9	3 1.	2 1.4	7.1
Average	34.3	21.7	2.0	14.9	146.1	70.0	14.1	0.81	226.1	. 9.9	12.40	1.77	57.9 1	8.87 4	42.6 l	7.9 6	4.6 (1.54	6.3	4.7 5	6.5	1.7	71 37.	0 10.	7 2.3	3 0.97	1.31	7.59
Stdev	5.3	4.4	0.0	2.8	46.2	12.9	1.9	0.11	52.7	1.7	1.54	2.85	14.4	5.28	7.5	5.1 3	3.7 (9.34	1.0	2.2		0.5	57 16.	.I 9.	4 0.8	3 0.17	0.68	0.52
JS-13-12A	50	18	1	11	06	60	11	0.5	180	Ŋ	9.10	8.34	82	27.1	59	15	79	0.8	\approx	4	4	30 0	9.	20	-	3 1.	l 0.2	5.3
JS-13-12B	40	19	1	13	113	80	11	0.6	213	9	4.20	3.21	40	38.6	30	14	9	0.3	$\overset{\circ}{2}$	5	7 2	80 0	.5	20	Ţ	4 1.	2 0.1	5.1
JS-13-12C	40	16	2	11	76	80	11	0.7	177	Ŋ	9.70	8.31	82	36.5	63	17	83	0.3	\$	З	5	30 <0	1.	59 •	Ţ	2 1.0	0.3	6.8
JS-13-13B	40	18	2	14	84	60	10	0.7	129	4	9.20	3.75	58	27.1	41	11	115	<0.2	\$	$\overset{<}{\sim}$	4	30 0	1.	10	Ţ	2 1.	5 0.3	9.8
Average	42.5	17.8	1.5	12.3	90.8	70.0	10.8	0.63	174.8	5.0	8.05	5.90 (55.5 3.	2.33 4	<i>t8.3</i> 1	4.3 7	0.8 (9.47 -		4.0 5	5.0	0.4	ŧ0 37.	3	2.8	3 1.20	0.23	6.75
Stdev	5.0	1.3	0.6	1.5	15.9	11.5	0.5	0.10	34.6	0.8	2.58	2.81	20.4	6.09 1	15.5	2.5 4	6.1 (9.29 -		I.0	.4	0.2	22.	3	1.0	0.22	0.10	2.17
JS-13-14A	40	21	2	14	96	60	13	1.0	97.3	3	9.90	2.87	34	14.3	28	17	18	<0.2	$\stackrel{<}{\sim}$	<3	ы V	30 <0	.1	\Im	7	3 0.5	3 0.2	7.9
JS-13-14B	90	27	2	18	112	90	17	1.2	164	S	14.4	4.16	67	33.1	64	28	70	<0.2	<3	<3	5	30 <0	.1	3	. 7	4 0.9	0.7	11.3
JS-13-14C	50	30	2	21	142	100	17	1.3	131	4	14.4	3.45	53	21.9	24	22	68	<0.2	\hat{c}	\$	2	30 <0		2	ŝ	4 1.	l 0.4	9.7
JS-13-14D	100	30	7	18	115	90	18	1.3	157	2	17.4	4.37	52	23.9	35	34	96	<0.2	<3	\$3	4	30 <0	.1	0	7	3 1.	1 0.4	13.6
JS-13-14E	60	29	7	20	137	80	17	1.3	179	2	17.2	4.79	63	27.9	41	26	42	<0.2	<3	\$3	5	30 <0	1.	10	3	3 1.	l 0.4	8.6
Average	68.0	27.4	2.0	18.2	120.4	84.0	16.4	1.22	145.7	4.4	i 4.66	3.93	53.8 2	4.22 3	38.4 2	5.4 5	8.8		1		1.2	1	1	- 7	4 3.4	1 1.00	0.42	10.22
Stdev	25.9	3.8	0.0	2.7	19.0	15.2	1.9	0.13	32.1	0.9	3.03	0.76	12.8	7.01 i	15.7	6.4 ź	9.8	-	-	-		-		0.	5 0.5	5 0.14	0.18	2.28

Table A2. Whole-rock analyses for trace and rare earth elements and selected ratios for the Smalls Falls Formation in Maine.

Table A2. Continued.

JS-13-10A 104 146 JS-13-10B 78.3 102 JS-13-10C 54.3 118 JS-13-10D 102 177																ΙD	2 C	LL.	Eu*
JS-13-10A 104 146 JS-13-10B 78.3 102 JS-13-10C 54.3 118 JS-13-10D 102 177						P.	mq												
JS-13-10B 78.3 102 JS-13-10C 54.3 118 JS-13-10D 102 177	436	25.4	33.0	70.9	8.43	30.6	6.00	1.20	5.23	0.84	4.91 (.95 2	.91 0.	45 2	.8 0.4	4 11.79	0.980	1.067	0.970
JS-13-10C 54.3 118 JS-13-10D 102 177	326	23.5	31.9	69.3	8.17	29.3	5.70	1.08	5.07	0.85	4.78 (.88 2	.68 0.	43 2	.6 0.4	0 12.27	0.989	1.069	0.900
JS-13-10D 102 177	210	14.4	19.3	41.7	4.92	17.8	3.50	0.67	3.32	0.53	2.92 (.51 1	.54 0.	25 1	5 0.2	3 12.87	7 0.986	1.065	0.905
	427	22.4	27.1	57.1	6.84	24.4	4.70	0.96	4.44	0.77	4.16 (.80 2	.49 0.	42 2	.6 0.4	0 10.42	2 0.967	1.080	0.940
JS-13-10E 129 209	540	24.7	36.3	76.8	9.06	32.6	6.30	1.27	5.47	06.0	4.77 (.92 2	.79 0.	44 2	.7 0.4	0 13.4	1 0.976	1.067	0.971
Average 93.5 150.4	387.8	22.08	29.52	63.16	7.48	26.94	5.24	1.04	4.71	0.78	1.31 0	.81 2.	48 0	10 2.4	4 0.3	7 12.16	0.980	1.070	0.937
Stdev 28.3 43.5	125.0	4.44	6.60	13.97	1.65	5.94	1.14	0.24	0.86	0.15 ().83 ()	.18 0.	55 0.1	8 0.5	3 0.0	3 1.15	0.009	0.006	0.034
15 13 11 VII 51 31	007	C 72	7 7 7	70 U	0 70	5 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	10	1 11	6 73	01 1	ניוי	0 0	30	3	2 0 1	2 11 0	0 057	1 007	0.051
IS-13-11R 129 278	557 557	30.8	34.0	7.07	9.09	33.3	6 70	1 53	2.06	96.0	235 1	0 7 C	30 o.			4 10 G	1000 F	1 089	1 099
S-13-11C 101 219	268	28.6	33.0	71.2	8.57	31.3	6.10	1.40	6.54	0.97	5.52	.11	.19 0.	51 3	0 0.5	1 11.00	0.975	1.070	1.067
JS-13-11D 130 330	417	25.5	39.4	79.9	9.21	32.9	6.10	1.56	5.84	0.88	4.94 i	.02 2	.91 0.	49 2	8 0.4	8 14.07	7 0.968	1.059	1.228
JS-13-11E 158 265	604	32.6	39.1	84.9	10.0	36.9	7.20	1.77	7.35	1.12	6.34	.31 3	.62 0.	50 3	4 0.5	9 11.50	0.989	1.053	1.151
JS-13-11F 147 185	648	32.9	43.2	93.8	11.0	40.3	7.60	1.80	7.70	1.15	6.36	.31 3	.58 0.	51 3	.3 0.6	0 13.09	0.992	1.055	1.120
JS-13-11G 164 201	801	37.8	48.9	107	12.3	45.2	8.80	2.12	8.82	1.27	7.30	.48 4	.04 0.	55 3	.7 0.6	4 13.27	2 1.006	1.043	1.157
Average 135.0 251.0	531.3	31.77	39.14	84.00	9.99	36.49	7.09	1.66	6.99	1.06	5.99 I	.24 3.	44 0.	6 3.2	1 0.5	5 12.18	0.977	1.066	1.110
Stdev 22.7 51.4	175.6	3.97	5.53	12.68	1.28	4.86	0.94	0.25	1.05	0.13 (0.79 0	.15 0.	36 0.1	0.2	9 0.0	5 1.29	0.020	0.019	0.087
JS-13-12A 137 47	466	16.6	14.7	32.7	3.98	14.7	3.00	0.72	3.24	0.50	3.12 (.66 1	.85 0.	32 1	9 0.3	4 7.7	1 0.983	1.070	1.097
JS-13-12B 141 63.8	615	20.3	15.6	33.5	4.08	15.5	3.50	0.89	3.84	0.62	3.75 (.78 2	.27 0.	36 2	.2 0.3	5 7.09	90.967	1.056	1.137
JS-13-12C 108 148	216	22.0	11.7	25.7	3.20	11.9	2.60	0.96	3.17	0.63	3.84 (.81 2	.53 0.	39 2	5 0.3	8 4.68	3 0.966	1.079	1.454
JS-13-13B 202 230	187	20.5	29.7	64.0	7.48	26.9	5.50	1.15	4.65	0.78	4.10 (.74 2	.31 0.	35 2	.3 0.3	4 12.9	1 0.990	1.063	1.010
Average 147.0 122.2	371.0	19.85	17.93	38.98	4.69	17.25	3.65	0.93	3.73	0.63 、	3.70 0	.75 2.	24 0	86 2.2	3 0.3	5 8.11	0.976	1.067	1.175
Stdev 39.5 84.4	205.3	2.30	8.03	17.05	1.90	6.62	1.29	0.18	0.69	0.11 (0.42 0	.07 0.	28 0.1	3 0.2	5 0.0	2 3.46	0.012	0.010	0.194
IS-13-14A 150 128	365	13.8	21.8	47.0	5.37	18.6	3.40	0.81	2.72	0.48	2.58 (.47 1	.58 0.	25 1	5 0.2	3 14.5	3 1.002	1.069	1.152
JS-13-14B 178 181	377	31.5	37.2	78.0	9.39	33.9	6.80	1.44	5.93	1.01	5.86	14 3	.58 0.	59 3	6 0.5	7 10.3	3 0.962	1.077	1.007
JS-13-14C 215 136	515	26.1	38.9	83.2	9.88	34.6	6.50	1.32	5.32	0.87	4.85 (.96 3	.00 00.	51 3	.2 0.5	0 12.10	5 0.978	1.085	0.998
JS-13-14D 221 131	447	45.6	48.5	104	12.2	43.3	8.40	1.67	6.95	1.27	7.99	.58 5	.17 0.	85 5	.3 0.8	3 9.15	5 0.986	1.072	0.940
JS-13-14E 221 148	524	31.6	31.0	68.1	8.12	28.4	5.90	1.32	5.13	0.95	5.56	18 3	.0 69.	50 3	5 0.5	8.8	5 0.988	1.088	1.035
Average 197.0 144.8	445.6	29.72	35.48	76.06	8.99	31.76	6.20	1.31	5.21	0.92	5.37 1	.07 3.	40 0.	6 3.4	2 0.5	4 11.01	0.983	1.078	1.027
Stdev 31.8 21.6	74.4	11.46	9.89	20.88	2.51	9.09	1.82	0.31	1.56	0.29	1.95 0	40 1.	30 0.	22 1.3	5 0.2	1 2.360	0.015	0.008	0.078
Notes: Elements not detecte (SN) using data for PAAS (7	d and low Tavlor an	ver limit d McLei	s of det∢ nnan 19	sction (ii 85): Ce/	n paren 'Ce* _{sn =}	theses): = 2Ce _{sn} /	Be (5 p '(Lasn 1	pm); A - Pr _{sv});	g (1 pp ¹ Eu/Eu*	m); In ($^{+}_{ev} = 2I$	(0.2 pp 3usn/(S	m). RE m _{en} +	E anon Gd _{en}).	ialies ar	e calcu	lated by	shale no	ormaliza	ion